

Chemquest 12 Quantum Numbers Answers



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38 Critical Thinking Questions 1. Using the quantum numbers (n, l, m_l, m_s) explain why each of the following is not an allowed combination of quantum numbers. The first one is done for you. a) $(3, 4, 1, +\frac{1}{2})$ Here $n = 3$ and so l can only have values of 0, 1, or 2. Here, however, l has a value of 4, which is impossible.

38 Critical Thinking Ques - Yumpu

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QUANTUM NUMBERS WORKSHEET 1. State the four quantum numbers, then explain the possible values they may have and what they actually represent. n - Principal Quantum Number: represents the energy level the electron is in, linked to the

QUANTUM NUMBERS WORKSHEET answers

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